

WATER-THIN EMULSIONS WITH LOW EMULSIFIER LEVELS

Field of the Invention

5 The invention relates to cosmetic and pharmaceutical formulations. More specifically, the invention relates to cosmetic and pharmaceutical formulations containing low levels of emulsifiers.

Background of the Invention

10 One of the most common vehicles for cosmetic and pharmaceutical products is the emulsion. Because they are formed by the dispersion of an oil in water, or water in an oil, they provide great versatility in the delivery of different types of active ingredients. A single oil and water formulation can be used to deliver both oil soluble and water soluble active components, thereby giving the formulation a range of potential activity that cannot be matched by a single phase system.

15 There are of course limitations to an emulsion vehicle, by virtue of its combination of two inherently incompatible phases. First, an emulsion ordinarily will have a certain amount of innate viscosity; while not necessarily a problem *per se*, the thickness of the emulsion can prevent its use in certain types of products or packaging that require a less viscous texture. In addition, in order to maintain a stable dispersion, it is 20 ordinarily necessary to add to the formulations substantial amounts of emulsion stabilizers and/or emulsifiers. The necessity of addition of these materials not only adds cost to the final product, but also has an effect on the quality of the final product, by affecting the way the emulsion breaks, as well 25 as how it feels on the skin. Added stabilizers can add to the 30

viscosity of the emulsion, and certain emulsifiers can be irritating to the skin of some users.

There have been attempts to overcome some of these difficulties. One common approach is the use of high pressure homogenization techniques, in which a crude emulsion is passed through a high pressure homogenizer to yield a relatively thin emulsion. This technique can contribute to a reduction in the viscosity of the emulsion, and such emulsions have even been stated as being made with relatively low levels of emulsifiers.

5 However, the emulsifiers used in these situations are either of the type that are known to cause irritation, i.e., non-ionic ethoxylated emulsifiers, or amphoteric, lecithin-type (phospholipid) emulsifiers, which, being naturally-occurring products, are rather costly to use. In some cases, these

10 emulsions will still require an addition of emulsion stabilizers to maintain stability over long periods of time. There thus continues to be a need for a water-thin emulsion which employs

15 minimal levels of a non-irritating emulsifier.

20 Summary of the Invention

The present invention relates to water-thin emulsions prepared by high pressure homogenization, in the presence of a non-phospholipid, non-ethoxylated "pseudoemulsifier" having a chemical composition comprising at least one hydrophobic moiety, and at least one polar moiety, the size and/or the planar at least two of one or of both of the types of moieties. Although not ordinarily effective for use alone in maintaining stability of emulsions, the pseudoemulsifiers have been shown to be highly effective in maintaining the stability of these water-thin emulsion, even at very low levels, i.e., less than 1%, and in addition are very mild and non-irritating to the skin. The water-thin emulsions find a variety of uses as a base for both cosmetic and pharmaceutical products. The invention also

provides a method for producing a water-thin emulsion, comprising mixing oil and water phases in the presence of the pseudoemulsifier, and subjecting the mixture to high pressure homogenization.

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Brief Description of the Figures

Figures 1a and 1b are schematic illustrations of possible different arrangements of hydrophobic and hydrophilic moieties in the pseudoemulsifiers of the present invention.

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Figure 2 is a schematic illustration of a surfactin molecule. "AA" represents an amino acid.

Detailed Description of the Invention

15 The emulsions of the invention have substantially no viscosity, i.e., they exhibit approximately the consistency of water. The consistency of the emulsions is primarily due to their processing under high pressure homogenization. In brief, the emulsion is prepared, in accordance with art-recognized techniques, by forming a crude mixture of the oil and water phases, in the presence of the appropriate emulsifier (as defined in more detail below), and passing it through a high pressure homogenizer for a time sufficient to achieve a stable emulsion. The pressure sufficient to achieve the stable emulsion ranges 20 from about 15,000 to about 45,000 psi, or about 1000-3100 bar, preferably about 1300-3000 bar, utilizing one or more passes. Suitable homogenizers for this purpose are commercially 25 available; these include a microfluidizer, Dee Bee 2000 (BEE International) and Cavitator (Five Star International). The preferred emulsion is an oil-in-water emulsion.

30 While the homogenization process is important in obtaining the appropriate consistency of the emulsion, this aspect alone is not unique. The water-thin emulsions of the present invention

utilize a type of non-ethoxylated, non-phospholipid "pseudoemulsifier", i.e., a compound or compounds that are not traditionally considered or used as emulsifiers, and/or which, when used alone in a traditional emulsion, are ordinarily not capable of stabilizing the emulsion at the very low levels used in the present invention. The type of emulsifier used is a non-ethoxylated, non-phospholipid having a chemical composition containing both hydrophobic moieties and polar(or hydrophilic) moieties, but with an asymmetrical molecular arrangement of the moieties. By "asymmetrical" is meant that the different moieties are of different sizes(e.g., short chain vs. long chain) and/or shapes(e.g., straight chain vs. cyclic), and/or are arranged in different three-dimensional planes within the composition. Preferably, the pseudoemulsifier is a single compound in which there are at least two of either the hydrophilic or hydrophobic moieties. The moieties of a given type may be the same or different, but are preferably different from each other, e.g., a compound will preferably have at least two different hydrophilic moieties, and/or at least two different hydrophobic moieties. Hydrophobic moieties can be any primarily hydrocarbon moiety, including, but not limited to, C1-40 linear or branched, substituted or unsubstituted alkyl, cycloalkyl, alkylene, alkaryl, or aryl groups. The polar or hydrophilic moieties are, for example, hydroxyl, carboxyl, ester, or amide groups, or hydrocarbon moieties that are highly substituted with such polar groups, or combinations thereof. Preferably, the moieties of the same type in a compound are also unequal in size or shape, for example, the hydrophobic moieties can be an alkyl and an aryl group, or two alkyls of different chain length. It is most preferable that the pseudoemulsifier have at least one closed, rigid structure, which can be either hydrophilic or hydrophobic in nature, for example, an aliphatic ring with ether, ester or amide linkages, or an aromatic ring, the rigid structure being

anchored by at least one long-chain, i.e., C8-22, straight or branched hydrophobe or hydrophile, and one or two short chain hydrophiles or hydrophobes. Particularly preferred long-chain moieties are C8-22 fatty acid moieties, such as stearate or palmitate. Schematic illustrations of some possible different arrangements of groups in a single compound are shown in Figure 1. The preferred arrangements provide for a broadly dispersed hydrophilic domain separating the components of the hydrophobic domain. Ordinarily, such molecules will not be readily water-soluble or oil-soluble at room temperature, but will be readily dispersible in either at higher temperatures.

In the case in which the emulsifier does not have a rigid structure *per se*, it is possible to confer the necessary rigidity by combining the emulsifier with a polymer having dispersed hydrophilic groups along the molecule, to form an emulsifier system. Polymers of this type will hydrogen-bond within the system, thereby creating the structure needed to mimic the desirable structure described above. Examples of useful polymers of this type include sugars, such as disaccharides, e.g., sucrose, lactose, or maltose, and polysaccharides, e.g., cellulose, pectin, xanthan gum, or amylose; or a predominantly hydrophilic peptide or protein, i.e., ones having a preponderance of hydrophilic or polar amino acid residues.

Although it is preferred that the emulsifier components be combined in a single molecule, it is also possible to create a mixture of compounds, having a similar balance of polar and hydrophobic moieties and "asymmetry" as described above, i.e., comprising more than one compound, the compounds used containing a mixture of hydrophobic components and polar or hydrophilic components as described above for the moieties of a single compound, and which mixture will accomplish the same result as the use of the single compound. The combination of compounds should have an overall average HLB value of between 6 and 8. In

one embodiment, the components used can incorporate one hydrophilic and one hydrophobic moiety in a single molecule, for example, a glycerol ester, such as polyglyceryl-2-isostearate or a sucrose or glucose ester, such as sucrose stearate or sucrose cocoate, in combination with one or more compounds which have hydrophilic or hydrophobic moieties. As with the single compound embodiment, it is preferred that there be at least two hydrophobic moieties or at least two hydrophilic moieties present in the components employed. In the case in which separate compounds are used to contribute the individual hydrophilic and hydrophobic moieties, however, the requisite structure or rigidity will not be present without the addition of a polymer to tie the components together. Therefore, with separate compounds being used, the addition of a polymer with disperse hydrophilic groups is important; the polymer will act, as described above, by forming hydrogen bonds with the other components, forming a cohesive system comparable to the single compound system. When used, in either the single compound or multiple compound system, the polymer is employed in an amount of about 0.1 to about 2%. Although the mixture *per se* does not necessarily have an innate asymmetry, except perhaps in the different size and/or conformation of the different compounds, the combination *in situ* in the emulsion will self-assemble asymmetrically.

An example of one embodiment of the single compound structures is a group of anionic emulsifiers of the type that is disclosed in PCT Publication No. WO 91/01970, the contents of which are incorporated herein by reference. One class of compounds are 2-amidocarbonyl-benzoic acid surfactants having the formula (I):

wherein R₁ and R₂ are independently H or (CH₂)_nCH₃, wherein n=8-22, provided that at least one of R₁ and R₂ is H, wherein M⁺ is a cation selected from the group consisting of H, Na, K, NH₄ and derivatives thereof (for example, basic amino acids), Ba, Ca, Mg,

Al, Ti, and Zr, and y is an integer of a value satisfying the valency of M^+ . Particularly preferred among this class of surfactants is a monovalent salt of stearyl amidobenzoic acid, preferably a sodium salt, also known as RM1. This compound and others of its type are commercially available from Stepan Company, Northfield, Illinois. These compounds are known surfactants, which have previously been reported to form stable oil-in-water emulsions when combined with a low HLB emulsifier, or a polymeric emulsifier. However, in the present case, these emulsifiers can be used as the sole emulsifier, at very low levels (i.e., as low as 0.25%) to achieve a stable emulsion; this result is particularly unexpected with an anionic emulsifier alone or at low levels. As can readily be seen from the structure depicted above, these compounds contain two hydrophobic groups, in the presence of the aromatic ring and the long-chain fatty acid side chain, separated by two hydrophilic groups, namely, the carboxylate and amide portions of the molecule.

Another example of a compound meeting the requirements outlined above is a compound, or a group of structurally-related compounds, all known as surfactin. This material is naturally occurring, being produced by fermentation of certain strains of *Bacillus subtilis*, and is commercially available from Showa Denko, KK, Japan. The structure of the molecule is unusual, being composed of a large hydrophilic ring containing seven amino acids bonded to each other by six peptide bonds and an ester bond, and having on either side two short-chain hydrocarbons with free carboxyl groups carrying anionic charges, (the ring and carboxyl groups constituting the hydrophilic domain), with a hydrophobic domain comprising a long chain fatty acid residue. A schematic representation of this type of molecule is seen in Figure 2. It will readily be seen that these molecules possess an asymmetrical arrangement of the hydrophilic and hydrophobic

moieties, and also have the desired rigidity in the presence of the amino acid ring structure.

As a third example, and one in which there is not a single compound, is a combination of xanthan, polyglucomannan, a high 5 HLB emulsifier, and a low HLB emulsifier. Such a combination is available from Uniqema as part of the Arlatone Versaflex Series of high performance emulsion stabilization systems.

The water-thin emulsions of the present invention have two distinct advantages over prior art water-thin emulsions. The 10 first advantage is that the pseudoemulsifiers of this type are innately mild, being relatively non-water soluble, and hence, non-reactive with skin, and therefore are less irritating by nature than an ethoxylated emulsifiers. In addition, the compounds of these systems, even though not ordinarily effective 15 as emulsifiers on their own, have proven to be unusually effective in stabilizing this type of emulsion, thereby improving even further the mildness of the emulsions by reducing the amount of emulsifier needed. The oil-in water emulsions of the invention ordinarily will contain no more than about 3% total 20 pseudoemulsifier, preferably no more than 2% pseudoemulsifiers, and more preferably, no more than 0.5% pseudoemulsifier. Because of the unusual properties of these pseudoemulsifiers, the emulsion is stable even in the substantial absence of added 25 emulsion stabilizers. It may, however, be desired to thicken slightly the water-thin emulsion depending on the desired nature of the final product. Therefore, it is possible to add to the emulsion a small amount of one or more cosmetic powders, not for 30 stabilization, but merely to modify the viscosity of the product. Examples of types of powders that can be used in the present emulsion are silica powders, polymethylmethacrylate, bismuth oxychloride, boron nitride, barium sulfate, mica, sericite, muscovite, synthetic mica, titanium oxide coated mica, titanium oxide coated bismuth oxychloride, talc, polyethylene, nylon,

polypropylene, acrylates/alkyl acrylates crosspolymer, acrylamide copolymers, and the like. The powders can be used in an amount of up to about 20%, but ordinarily the powders will be used in small amounts, generally no greater than about 5% of the total weight 5 of the emulsion, more preferably no greater than 2%. In certain embodiments, the emulsion will contain less than 0.5% by weight of powders.

In order to prepare the water-thin emulsions of the invention, the pseudoemulsifier is combined with any standard oil 10 and water emulsion components. The aqueous phase may be any cosmetically acceptable water based material, such as deionized water, or a floral water. The oil phase may be any cosmetically or pharmaceutically acceptable oil, such an oil being defined for the present purpose as any pharmaceutically or cosmetically 15 acceptable material which is substantially insoluble in water. As the oils can perform different functions in the composition, the specific choice is dependent on the purpose for which it is intended. The oils may be volatile or non-volatile, or a mixture 20 of both. For example, suitable volatile oils include, but are not limited to, both cyclic and linear silicones, such as octamethylcyclotetrasiloxane and decamethylcyclopentasiloxane; or straight or branched chain hydrocarbons having from 8-20 carbon atoms, such as decane, dodecane, tridecane, tetradecane, and C8-25 20 isoparaffins.

Non-volatile oils include, but are not limited to, 25 vegetable oils, such as coconut oil, jojoba oil, corn oil, sunflower oil, palm oil, soybean oil; carboxylic acid esters such as isostearyl neopentanoate, cetyl octanoate, cetyl ricinoleate, octyl palmitate, dioctyl malate, coco-dicaprlylate/caprate, decyl 30 isostearate, myristyl myristate; animal oils such as lanolin and lanolin derivatives, tallow, mink oil or cholesterol; glyceryl esters, such as glyceryl stearate, glyceryl dioleate, glyceryl distearate, glyceryl linoleate, glyceryl myristate; non-volatile

silicones, such as dimethicone, dimethiconol, dimethicone copolyol, phenyl trimethicone, methicone, simethicone; and nonvolatile hydrocarbons, such as isoparaffins, squalane, or petrolatum.

5 The ratio of oil phase:water phase in the emulsion is not critical, and can range from about 10:90 to about 50:50, but is more preferably from about 30:70 to about 40:60.

10 The pseudoemulsifier is ordinarily added to the phase in which it is soluble, or to either phase if it is not soluble in either, along with any active components which may be desired in the emulsion, and all components mixed together at low pressure. The mixture is then subjected to high pressure mixing. By "high pressure" in the present context is meant a pressure of at least about 15,000 psi, preferably at least about 25,000 psi, more 15 preferably about 35,000 psi; generally, a single pass through the high pressure equipment is adequate to achieve an emulsion of the desired type at higher pressures, although at lower pressures, more than one pass may be required. The amount of pseudoemulsifier employed is preferably no more than 2%, more 20 preferably no more than 1%. Amounts of as low as 0.25% can be employed, although about 0.5 to about 1% is generally preferred. It will be understood that the amount of pseudoemulsifier and 25 level of pressure can be varied inversely, with a higher pressure treatment allowing the use of lower levels of emulsifier to produce a stable emulsion, while lower pressure treatments will ordinarily require a level of emulsifier at the higher end of the effective range. The particle size distribution is normally narrow, and very small, usually in the range of about 50-150 nm, 30 preferably with an average size of about 50-100 nm, more preferably about 50 nm.

An oil-in-water emulsion so prepared can be used as such, or it can be further used as a base to which an additional water phase, particularly one enriched with active ingredients that may

be too temperature sensitive to be added to the high-temperature pre-mix, can be added, under low pressure. This approach yields an oil-in-water emulsion in the form of either a spray, lotion or cream. In the preparation of such a composition, the oil-in-water emulsion can be added to the water in an amount ranging 5 from about 90:10 emulsion :water to 10:90 emulsion:water.

If desired, the viscosity of the resulting products can be increased by the addition of water-soluble thickeners such as 10 acrylates crosspolymers and copolymers, carbomer, guar gum, carageenan, cellulosics, mannan, sulfonic acid polymers, acrylamide copolymer, xanthan gum and the like. Preferably, the amount of thickener ranges between about .01 to about 1%, preferably no more than about 0.5%.

In a particularly preferred embodiment, the water-thin 15 emulsion is added to a water-in-oil emulsion, so as to prepare a multiple phase emulsion. This type of emulsion is valuable for a number of reasons. First, it provides a means for incorporating actives in the same vehicle which would not ordinarily be compatible in the same phase, by incorporating them in different 20 phases. It also is a useful vehicle for delayed release of actives on and into the skin, by virtue of the necessity of passing through the multiple phases. Despite their clear value; however, such emulsions are not frequently employed, as the 25 additional phase introduces further problems with stability, and therefore, they frequently require the use of large quantities of emulsifiers and/or emulsion stabilizers. It has now been found, unexpectedly, that the water-thin emulsion can provide a basis for the preparation of a multiple emulsion, serving as the outer water phase, without the need for large quantities of 30 emulsifiers. In such preparation, the premade water-thin emulsion serves as the water phase, and is mixed, under normal, low-pressure conditions, with a premade standard water-in-oil emulsion. The two emulsions are preferably combined in a ratio

of about 80 water-thin emulsion:20 water-in-oil emulsion up to 50:50 water-thin:water-in-oil, to yield a stable multiple emulsion. Surprisingly, these multiple emulsions can be prepared with no more than 2% emulsifiers, and preferably no more than 5 about 1.5% emulsifiers total in the multiple emulsion.

Even more unexpectedly, the multiple emulsions can be prepared with an even number of phases, e.g., four phases, rather than the standard uneven number ordinarily found in multiple emulsions, such as water-in-oil-in-water. This is made possible 10 by the small droplet size of the oils in the water-thin emulsion, which essentially presents itself as water to a standard emulsion, and is therefore readily incorporated without the addition of large amounts of emulsifiers.

Generally speaking, the multiple emulsion can be prepared 15 with little or no "traditional" emulsifier, a traditional emulsifiers being one which, unlike the pseudoemulsifiers, are capable of stabilizing emulsions on their own, even at relatively low levels. When combined with a water-thin oil-in-water emulsion to make the multiple emulsion, the multiple emulsion may 20 employ small amounts of a traditional oil-in-water emulsifier. Examples of useful oil-in-water emulsifiers include, but are not limited to, sorbitol derivatives, such as sorbitan monolaurate and polysorbate 20; ethoxylated alcohols such as laureth-23, 25 ethoxylated fatty acids such as PEG-1000 stearate; amidoamine derivatives such as stearamidoethyl diethylamine; sulfate esters such as sodium lauryl sulfate; phosphate esters such as DEA cetyl phosphate; fatty acid amine salts such as TEA stearate; and mixtures thereof. Additional examples can be found in McCutcheon's Emulsifiers and Detergents, 2000, the contents of 30 which are incorporated herein by reference. If used, this type of emulsifier is incorporated in quantities of no more than about 2% by weight of the multiple emulsion, preferably no more than 1%, and more preferably, no more than about .5%. Stabilizers or

thickeners, if used at all, can be employed as described for the water-thin emulsion alone.

The compositions of the invention can be used for any cosmetic or pharmaceutical purpose in which an standard or 5 multiple emulsion is normally useful. For cosmetic purposes, the emulsions can be used in makeup products as well as skin-care products. In such cases, it may be desirable to incorporate into the emulsion additional components usually associated with the desired cosmetic uses, such as additional preservatives, 10 fragrances, emollients, antiseptics, antiinflammatories, antibacterials, stabilizers, sunscreens, antioxidants, vitamins, pigments, dyes, humectants, and propellants, as well as other classes of materials the presence of which in the compositions 15 may be cosmetically, medicinally, or otherwise desired. Such components can be found in the CTFA International Cosmetics Ingredients Dictionary, *supra*.

For pharmaceutical or therapeutic cosmetic use, the emulsion can incorporate any variety of topically applied therapeutic agents, particularly those that will benefit from a 20 delayed release of active agents. Examples include, but are not limited to, agents for the eradication of age spots, keratoses and wrinkles, analgesics, anesthetics, anti-acne agents, antibacterials, antiyeast agents, antifungal agents, antiviral agents, 25 antidandruff agents, antidermatitis agents, antipruritic agents, antiemetics, antimotion sickness agents, anti-inflammatory agents, antihyperkeratolytic agents, anti-dry skin agents, antiperspirants, antipsoriatic agents, antiseborrheic agents, hair conditioners and hair treatment agents, antiaging 30 agents, antiwrinkle agents, antiasthmatic agents and bronchodilators, sunscreen agents, antihistamine agents, skin lightening agents, depigmenting agents, vitamins, corticosteroids, self-tanning agents, hormones, retinoids, such as retinoic acid, 13-cis retinoic acid, and retinol, topical

cardiovascular agents, clotrimazole, ketoconazole, miconazole, griseofulvin, hydroxyzine, diphenhydramine, pramoxine, lidocaine, procaine, mepivacaine, monobenzone, erythromycin, tetracycline, clindamycin, mecloxyline, hydroquinone, minocycline, naproxen, 5 ibuprofen, theophylline, cromolyn, albuterol, topical steroids such as hydrocortisone, hydrocortisone 21-acetate, hydrocortisone 17-valerate, and hydrocortisone 17-butyrate, betamethasone valerate, betamethasone dipropionate, triamcinolone acetonide, fluocinonide, clobetasol, propionate, benzoyl peroxide, 10 crotamiton, propranolol, promethazine, vitamin A palmitate, vitamin E acetate and mixtures thereof.

The invention is further illustrated by the following non-limiting examples.

15 Examples

Example 1

20 This example illustrates the preparation of an oil-in-water emulsion of the invention.

	Material	Weight %		
		Batch 1	Batch 2	Batch 3
25 Water Phase				
	Deionized water	65.25	64.70	62.50
	Disodium EDTA	0.10	0.10	0.10
	Methyl paraben	0.05	0.05	0.05
	Butylene glycol	3.00	3.00	3.00
30	phenoxyethanol	0.40	0.40	0.40
Oil Phase				
	Behenyl alcohol	0.75	0.75	0.75

Pentaerythrityl tetraoctanoate	30.00	30.00	-----
Stepan RM1	0.50	1.00	0.50
C12-15 alkyl benzoate	-----	-----	30.00

5 The water phase materials are heated to 85-90°C. The oil phase materials are heated to 85°C. The oil phase materials are added to the water phase materials using a Silverson (low pressure) homogenizer. The batch is then run through a DEE/BEE 2000 homogenizer at 40,000 psi, and cooled to room temperature in
10 the machine.

Example 2

This example illustrates the use of the emulsions prepared above in creating a variety of low emulsifier skin care products.

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A. Low emulsifier creme

Material	Weight %
Phase I	
20 Deionized Water	19.00
acrylamide copolymer(1.5%)	10.00
Glycerine	10.00
antiinflammatory polysaccharide	1.00
Phase II	
25 Batch 3	39.60
Acrylates/C10-30 alkyl acrylates	
Crosspolymer(2%)	20.00
triethanolamine	0.40

30

Phase I and Phase II components are separately premixed by propeller mixing. The two phases are then mixed to homogeneity with a propeller or paddle.

5 B. Low emulsifier milk lotions

Material	Weight %
Batch 2	97.00 99.00
Magnesium ascorbyl phosphate	3.00 -----
10 N-acetyl glucosamine	----- 1.00

C. Low emulsifier cleanser

Material	Weight %
Phase I	
15 Deionized water	18.70
methyl paraben	0.10
Phase II	
glycerine	10.00
20 Phase III	
Acrylamide copolymer (1.5%)	10.00
Phase IV	
25 Acrylates/C10-30 alkyl acrylates crosspolymer (2%)	20.00
Phase V	
30 Batch 2	39.60
Phase VI	
triethanolamine	0.40
deionized water	0.50

Phase VII	
Germall 115	0.20
deionized water	0.50

5 Phase I materials are heated to 75°C, and cooled to room temperature. Phases II, III, and IV are added to Phase I under propeller agitation. After addition of Phase IV, viscosity increases, requiring a change to a paddle. Phases V, VI and VII are then added to the mixture, and mixed to homogeneity.

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Example 3

15 This example illustrates the process of preparing a multiple emulsion according to the invention.

A. A water-in-oil phase is prepared as follows:

	Material	Weight %
Phase I		
	Cyclomethicone/dimethicone	5.00
	Phenyltrimethicone	5.00
	Dimethicone/copolyol crosspolymer	7.00
25	Cyclomethicone	1.00
	Dimethicone	8.00
Phase II		
	Xanthan gum	0.20
30	Deionized water	64.30
	Sodium chloride	1.00
	Butylene glycol	5.00
	Parabens	0.50

The oil phase ingredients are combined together, and the water phase ingredients are combined together. The water phase is then slowly added to the oil phase, and homogenized until uniform.

5 B. Water-thin, low emulsifier emulsion

	Material	Weight %
Phase I		
10	deionized water	32.50
	Arlatone Versaflex High	
	Performance Emulsion Stabilizer*	1.00
Phase II		
15	Deionized water	32.05
	Methyl paraben	0.20
	Butylene glycol	3.00
	Phenoxyethanol	0.40
Phase III		
20	Behenyl alcohol	0.75
	Pentaerythrityl tetraethylhexanoate	30.00
	Beta-carotene	0.10
25	*Uniqema	

In Phase I, the emulsifier is added to water at 80°C. Phase II ingredients are added to Phase I at 80°C. Phase III ingredients are combined and then homomixed with Phase I and II ingredients at greater than 10,000 rpm for 5 minutes. The combined components are then passed through a microfluidizer at 16,000 psi three times to achieve a water-thin emulsion.

C. Multiple emulsion

Material	Weight %
5	
Polysorbate 20	0.20
Carbopol	1.00
O/W emulsion from B.	78.80
W/O emulsion from A.	20.00

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The O/W emulsion is combined with the Carbopol using static mixing. Polysorbate 20 is then added. The W/O emulsion is slowly added to the O/W phase utilizing static mixing. When the addition is complete, the mixing is continued for about 5 minutes 15 until the multiple emulsion is uniform.

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Similar positive results are obtained in preparing a multiple emulsion as described above, utilizing RM1 as the pseudoemulsifier, and combining the O/W emulsion and the W/O emulsion in a ratio of 60:40.

What is claimed is:

1. A water-thin emulsion comprising a non-phospholipid, non-ethoxylated pseudoemulsifier system, the system having a chemical composition with at least one hydrophobic moiety and at least one polar moiety, the size, shape and/or planar arrangement of the hydrophobic and polar moieties being asymmetrical with respect to each other.

10 2. The emulsion of claim 1 in which the system is a single compound having at least two hydrophobic moieties, at least two polar moieties, or at least two of both hydrophobic and polar moieties.

15 3. The emulsion of claim 1 in which the system is a mixture of compounds comprising at least two hydrophobic moieties, at least two polar moieties, or at least two of both hydrophobic and polar moieties.

20 4. The emulsion of claim 1 which is substantially free of phospholipid and ethoxylated emulsifiers.

5. The emulsion of claim 1 which contains no more than 2% by weight of the pseudoemulsifier.

25 6. The emulsion of claim 1 which contains no more than 1% of the pseudoemulsifier.

30 7. The emulsion of claim 1 in which the pseudoemulsifier is a 2-amidocarbonyl-benzoic acid compound having the formula (I) wherein R₁ and R₂ are independently H or (CH₂)_nCH₃, wherein n=8-22, provided that at least one of R₁ and R₂ is H, wherein M⁺ is a cation selected from the group consisting of H, Na, K, NH₄ and

derivatives thereof (for example, basic amino acids), Ba, Ca, Mg, Al, Ti, and Zr, and y is an integer of a value satisfying the valency of M^+ .

5 8. The emulsion of claim 7 in which the pseudoemulsifier is a monovalent salt of stearyl amidobenzoic acid.

9. The emulsion of claim 1 in which the pseudoemulsifier is surfactin.

10 10. The emulsion of claim 1 in which the system also comprises a polymer.

15 11. The emulsion of claim 10 in which the polymer comprises dispersed hydrophilic moieties.

12. The emulsion of claim 11 in which the polymer is selected from the group consisting of disaccharides, polysaccharides, or a predominantly hydrophilic peptide or protein.

20 13. The emulsion of claim 3 in which the system comprises at least one compound selected from the group consisting of glycerol esters, sucrose esters and glucose esters.

25 14. The emulsion of claim 13 in which the system comprises both a glycerol ester and a sucrose or glucose ester.

15. The emulsion of claim 14 in which the system also comprises a polymer.

30 16. The emulsion of claim 15 in which the polymer is selected from the group consisting of disaccharides, polysaccharides, and predominantly hydrophilic proteins or peptides.

17. The emulsion of claim 3 in which the system comprises xanthan, polyglucomannan, a high HLB emulsifier, and a low HLB emulsifier.

5 18. The emulsion of claim 13 in which the system comprises xanthan, polyglucomannan, a high HLB emulsifier, and a low HLB emulsifier.

10 19. A water-thin oil-in-water emulsion comprising a non-phospholipid, non-ethoxylated pseudoemulsifier system, the system having a chemical composition with at least two hydrophobic moieties, at least two polar moieties, or at least two of both hydrophobic and polar moieties, the size, shape and/or planar arrangement of the hydrophobic and polar moieties being 15 asymmetrical with respect to each other, each polar moiety being of a different size or shape than the other polar moiety if present, and each hydrophobic moiety being of different size or shape than the other if present.

20 20. The emulsion of claim 19 in which the hydrophobic moieties are of different chain lengths.

25 21. The emulsion of claim 19 in which at least one of the moieties has a closed ring structure.

22. The emulsion of claim 19 in which at least one of the moieties is a long straight-chain moiety.

30 23. The emulsion of claim 19 in which at least one of the moieties has a closed ring structure, and one of the moieties is a long, straight chain moiety.

24. The emulsion of claim 23 in which the system comprises a hydrophobic closed ring structure, and a long chain hydrophobe, separated from each other by a hydrophilic moiety.

5 25. The emulsion of claim 24 in which the hydrophilic moiety is selected from the group consisting of hydroxyl, amide, ester, or carboxyl moieties, hydrocarbons chains substituted with hydroxyl, amide, ester, or carboxyl moieties, and combinations thereof.

10 26. The emulsion of claim 23 in which the system comprises a hydrophilic closed ring structure, at least one carboxyl moiety, and a long chain fatty acid moiety.

15 27. The emulsion of claim 19 in which the emulsifier system comprises more than one compound.

28. The emulsion of claim 27 in which at least one of the compounds comprises a long, straight-chain hydrocarbon moiety.

20 29. The emulsion of claim 28 in which at least one of the compounds comprises a hydrophilic moiety selected from the group consisting of hydroxyl, amide, ester, or carboxyl moieties, hydrocarbons chains substituted with hydroxyl, amide, ester, or carboxyl moieties, and combinations thereof.

25 30. The emulsion of claim 29 in which the system further comprises a polymer selected from the group consisting of disaccharides, polysaccharides, and predominantly hydrophilic proteins or peptides.

30 31. A water-thin oil-in-water emulsion prepared by processing the combined oil and water phases comprising a non-phospholipid, non-ethoxylated pseudoemulsifier system, the system having a

chemical composition with at least one hydrophobic moiety and at least one polar moiety, the size, shape and/or planar arrangement of the hydrophobic and polar moieties being asymmetrical with respect to each other, through a high pressure homogenizer at a

5 pressure of at least about 15,000 psi.

32. The emulsion of claim 31 in which the system is a single compound having at least two hydrophobic moieties, at least two polar moieties, or at least two of both hydrophobic and polar

10 moieties.

33. The emulsion of claim 31 in which the system is a mixture of compounds comprising at least two hydrophobic moieties, at least two polar moieties, or at least two of both hydrophobic and polar

15 moieties.

34. A multiple emulsion comprising the emulsion of claim 1.

35. A multiple emulsion incorporating the emulsion of claim 6.

20 36. A multiple emulsion incorporating the emulsion of claim 19.

37. A multiple emulsion incorporating the emulsion of claim 27.

25 38. The emulsion of claim 33 that comprises no greater than 1% of traditional emulsifier.

39. The emulsion of claim 35 that comprises no greater than 1% of traditional emulsifier.

30 40. The emulsion of claim 36 that comprises no greater than 1% of traditional emulsifier.

41. The emulsion of claim 37 that comprises no greater than 1% of traditional emulsifier.

5 42. A multiple emulsion prepared by combining a water-in-oil emulsion with the emulsion of claim 1, and mixing to substantial homogeneity.

10 43. A method of making a water-thin, oil-in-water emulsion comprising processing the combined oil and water phases comprising a non-phospholipid, non-ethoxylated pseudoemulsifier system, the system having a chemical composition with at least one hydrophobic moiety and at least one polar moiety, the size, shape and/or planar arrangement of the hydrophobic and polar 15 moieties being asymmetrical with respect to each other, through a high pressure homogenizer at a pressure of at least about 15,000 psi.

**(19) World Intellectual Property Organization
International Bureau**



(43) International Publication Date
6 December 2001 (06.12.2001)

PCT

(10) International Publication Number
WO 01/91722 A3

(51) International Patent Classification⁷: A61K 7/00 (74) Agents: TSEVDOS, Estelle, J. et al.; Kenyon & Kenyon, One Broadway, New York, NY 10004 (US).

(21) International Application Number: PCT/US01/16952

(81) Designated States (national): AU, CA, JP.

(25) Filing Language: English

(84) Designated States (regional): European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR).

(30) Priority Data: 09/580 743 26 May 2000 (26.05.2000) US

Published:

(71) **Applicant: COLOR ACCESS, INC. [US/US]; 7 Corporate Center Drive, Melville, NY 11747 (US)**

- before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments

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(88) Date of publication of the international search report:
10 May 2002

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

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WO 01/91722 A3

(54) Title: WATER-THIN EMULSIONS WITH LOW EMULSIFIER LEVELS

(57) Abstract: The invention relates to a water-thin emulsion comprising a non-phospholipid, non-ethoxylated pseudoemulsifier system, the system having a chemical composition with at least one hydrophobic moiety and at least one polar moiety, the size, shape and/or planar arrangement of the hydrophobic and polar moieties being asymmetrical with respect to each other. The emulsion of the invention is prepared by high-pressure homogenization of a crude oil and water emulsion containing the pseudoemulsifier. The emulsions require little or no traditional emulsifier to maintain stability, and are particularly useful in the preparation of multiple emulsions.

INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 01/16952

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 A61K7/00

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

CHEM ABS Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 91 01970 A (STEPAN CO. ET AL.) 21 February 1991 (1991-02-21) cited in the application claims 1,31,34-47; example J ---	1,7,19, 31
X	WO 99 62482 A (SHOWA DENKO K.K.) 9 December 1999 (1999-12-09) claims 1,11,15; example 4 ---	1,9,19, 31
X	DE 198 46 772 A (BEIERSDORF AG) 13 April 2000 (2000-04-13) claim 1; example 1 ---	1
P, X	WO 00 33806 A (IMPERIAL CHEMICAL IND. PLC) 15 June 2000 (2000-06-15) claim 1; example 7.1 ---	1,3,17, 18 -/-



Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

* Special categories of cited documents :

- *A* document defining the general state of the art which is not considered to be of particular relevance
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- *O* document referring to an oral disclosure, use, exhibition or other means
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- *&* document member of the same patent family

Date of the actual completion of the international search

5 March 2002

Date of mailing of the international search report

11/03/2002

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3

INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 01/16952

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	CA 2 188 331 A (FMC CORP.) 20 April 1997 (1997-04-20) example 8.E -----	1, 3, 17, 18
A	M. DELEU ET AL.: "Structure, interfacial properties and functional qualities in foams and emulsions of surfactin..." SPEC. PUB. R. SOC. CHIM., 1999, pages 296-304, XP008001035 the whole document -----	9

FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

Continuation of Box I.2

Claims Nos.: 1,19,31,43 (all partial), 38-41

The term "pseudoemulsifier", appearing in independent claims 1, 19, 31 and 43 is not clear.

The term "traditional emulsifier" appearing in claims 38-41 is not understood.

Explanations given throughout the description about the above-mentioned terms and emulsions in general do not add clarity to the subject matter of these claims.

Moreover, the term "water-thin" appearing in the claims 1, 19, 31 and 43 is also not clear. It seems to relate to a composition defined by reference to a desirable characteristic or property.

The claims cover all compositions having this characteristic or property, whereas the application provides support within the meaning of Article 6 PCT and/or disclosure within the meaning of Article 5 PCT for only a very limited number of such compositions. In the present case, the claims so lack support, and the application so lacks disclosure, that a meaningful search over the whole of the claimed scope is impossible. Independent of the above reasoning, the claims also lack clarity (Article 6 PCT). An attempt is made to define the composition by reference to a result to be achieved. Again, this lack of clarity in the present case is such as to render a meaningful search over the whole of the claimed scope impossible.

Consequently, the search has been carried out for those parts of the claims which appear to be clear, supported and disclosed, namely those parts relating to the emulsion of claim 1 where the pseudoemulsifier system is to be chosen among the compounds presented in claims 7,8,9,13,14 and 17.

Claims 38-41 have not been searched.

The applicant's attention is drawn to the fact that claims, or parts of claims, relating to inventions in respect of which no international search report has been established need not be the subject of an international preliminary examination (Rule 66.1(e) PCT). The applicant is advised that the EPO policy when acting as an International Preliminary Examining Authority is normally not to carry out a preliminary examination on matter which has not been searched. This is the case irrespective of whether or not the claims are amended following receipt of the search report or during any Chapter II procedure.

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 01/16952

Patent document cited in search report	Publication date	Patent family member(s)		Publication date
WO 9101970	A 21-02-1991	US 5015415 A		14-05-1991
		AU 6038290 A		11-03-1991
		CA 2019584 A1		08-02-1991
		DE 69025404 D1		28-03-1996
		DE 69025404 T2		04-07-1996
		WO 9101970 A2		21-02-1991
		EP 0486510 A1		27-05-1992
		EP 0705591 A2		10-04-1996
		EP 0688601 A1		27-12-1995
		ES 2085353 T3		01-06-1996
		HK 1006839 A1		19-03-1999
		JP 4506538 T		12-11-1992
		NO 920509 A		06-04-1992
		SG 47901 A1		17-04-1998
		WO 9200283 A2		09-01-1992
		AT 152444 T		15-05-1997
		CA 2019704 A1		27-12-1991
		CA 2085119 A1		28-12-1991
		DE 69125957 D1		05-06-1997
		DE 69125957 T2		02-01-1998
		DK 638072 T3		02-06-1997
		EP 0638072 A1		15-02-1995
		ES 2103801 T3		01-10-1997
		JP 6501238 T		10-02-1994

WO 9962482	A 09-12-1999	AU 3956599 A		20-12-1999
		CN 1303267 T		11-07-2001
		EP 1082099 A1		14-03-2001
		WO 9962482 A1		09-12-1999
		JP 2000327591 A		28-11-2000

DE 19846772	A 13-04-2000	DE 19846772 A1		13-04-2000
		EP 0998910 A1		10-05-2000

WO 0033806	A 15-06-2000	AU 1288600 A		26-06-2000
		BR 9915963 A		21-08-2001
		CN 1333676 T		30-01-2002
		EP 1137396 A1		04-10-2001
		WO 0033806 A1		15-06-2000

CA 2188331	A 20-04-1997	CA 2188331 A1		20-04-1997
		US 6048564 A		11-04-2000
